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Key indicators

Single-crystal X-ray study
 $T = 290$ K
Mean $\sigma(\text{C}-\text{C}) = 0.003$ Å
 R factor = 0.020
 wR factor = 0.054
Data-to-parameter ratio = 22.5For details of how these key indicators were
automatically derived from the article, see
<http://journals.iucr.org/e>.Diaquabis(nicotinato- κO)bis(1,10-phenanthroline- $\kappa^2\text{N},\text{N}'$)lanthanum(III) trihydrate

The title compound, $[\text{La}(\text{C}_6\text{H}_4\text{NO}_2)_3(\text{C}_{12}\text{H}_8\text{N}_2)_2(\text{H}_2\text{O})_2] \cdot 3\text{H}_2\text{O}$, consists of a mononuclear complex and three non-coordinated water molecules. The La atom is coordinated by five O atoms of three nicotinate groups and two coordinated water molecules, and four N atoms of two bidentate phenanthroline molecules, in a tricapped trigonal-prismatic coordination geometry. The complex molecule and water molecules are linked into layers through $\text{O}-\text{H} \cdots \text{O}$ and $\text{O}-\text{H} \cdots \text{N}$ hydrogen bonds.

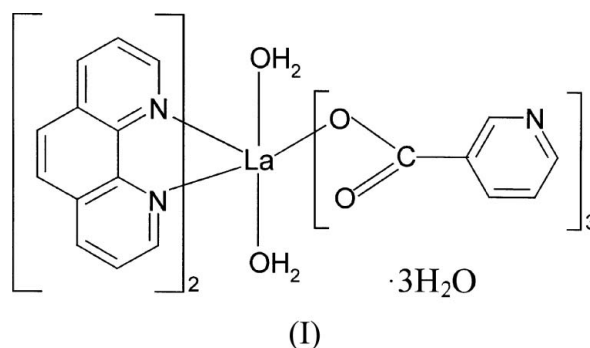
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Comment

The title compound, (I), is isostructural with its $[\text{M}(\text{C}_6\text{H}_4\text{NO}_2)_3(\text{C}_{12}\text{H}_8\text{N}_2)_2(\text{H}_2\text{O})_2] \cdot 3\text{H}_2\text{O}$ [$M = \text{Pr}$ (Yue *et al.*, 2004), Nd (Liu & Wang, 2004) and Ce (Liu *et al.*, 2005)] analogues. It consists of a mononuclear $[\text{La}(\text{C}_6\text{H}_4\text{NO}_2)_3(\text{C}_{12}\text{H}_8\text{N}_2)_2(\text{H}_2\text{O})_2]$ neutral molecule and three uncoordinated water molecules (Fig. 1). The La atom possesses trigonal prismatic coordination geometry (Table 1), in which the $\text{La}-\text{O}$ bond distances range from 2.4199 (10) to 2.5416 (10) Å and the $\text{La}-\text{N}$ bond distances range from 2.7565 (12) to 2.7757 (11) Å, which are similar to those in the previously reported isostructural complexes. In (I), the uncoordinated water molecules participate in the extensive hydrogen-bonding interactions formed between the water molecules and the N and O atoms of the nicotinate groups (Table 2), resulting in a layered structure.



Experimental

1,10-Phenanthroline monohydrate (0.0198 g, 0.10 mmol) and nicotinic acid (0.0123 g, 0.10 mmol) were completely dissolved in $\text{CH}_3\text{OH}/\text{H}_2\text{O}$ (30 ml, 1:1 *v/v*); $\text{La}(\text{CH}_3\text{COO})_3 \cdot 3\text{H}_2\text{O}$ (0.0158 g) was added and the mixture was stirred for 40 min. The resulting white slurry was filtered and the filtrate was left to stand at room temperature. Crystals of (I) suitable for X-ray analysis were obtained after two weeks (yield *ca* 35%).

Crystal data

$[\text{La}(\text{C}_6\text{H}_4\text{NO}_2)_3(\text{C}_{12}\text{H}_8\text{N}_2)_2 \cdot (\text{H}_2\text{O})_2] \cdot 3\text{H}_2\text{O}$
 $M_r = 955.70$
 Triclinic, $P\bar{1}$
 $a = 9.0676$ (1) Å
 $b = 12.9618$ (2) Å
 $c = 17.9351$ (3) Å
 $\alpha = 84.998$ (1)°
 $\beta = 80.800$ (1)°
 $\gamma = 84.190$ (1)°
 $V = 2064.73$ (5) Å³

$Z = 2$
 $D_x = 1.537$ Mg m⁻³
 Mo $K\alpha$ radiation
 Cell parameters from 985 reflections
 $\theta = 3.3\text{--}26.7^\circ$
 $\mu = 1.10$ mm⁻¹
 $T = 290$ (2) K
 Block, colorless
 $0.20 \times 0.20 \times 0.18$ mm

Data collection

Bruker SMART CCD area-detector diffractometer
 φ and ω scans
 Absorption correction: multi-scan (SADABS; Sheldrick, 1996)
 $T_{\min} = 0.791$, $T_{\max} = 0.811$
 45880 measured reflections

12414 independent reflections
 11284 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.020$
 $\theta_{\text{max}} = 30.7^\circ$
 $h = -12 \rightarrow 12$
 $k = -18 \rightarrow 17$
 $l = -25 \rightarrow 23$

Refinement

Refinement on F^2
 $R[F^2 > 2\sigma(F^2)] = 0.020$
 $wR(F^2) = 0.054$
 $S = 1.04$
 12414 reflections
 552 parameters
 H-atom parameters constrained

$w = 1/[\sigma^2(F_o^2) + (0.0318P)^2 + 0.2509P]$
 where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\text{max}} = 0.005$
 $\Delta\rho_{\text{max}} = 0.50$ e Å⁻³
 $\Delta\rho_{\text{min}} = -0.26$ e Å⁻³
 Extinction correction: SHELXL97
 Extinction coefficient: 0.0010 (2)

Table 1

Selected interatomic distances (Å).

La—O4	2.4199 (10)	La—N2	2.7565 (12)
La—O6	2.5200 (10)	La—N3	2.7566 (12)
La—O2	2.5355 (10)	La—N4	2.7678 (11)
La—O7	2.5367 (10)	La—N1	2.7757 (11)
La—O8	2.5416 (10)		

Table 2

Hydrogen-bond geometry (Å, °).

$D\text{---}H\cdots A$	$D\text{---}H$	$H\cdots A$	$D\cdots A$	$D\text{---}H\cdots A$
O7—H7A \cdots O1 ⁱ	0.85	1.77	2.612 (2)	172
O7—H7B \cdots O9 ⁱⁱ	0.85	1.85	2.701 (2)	178
O8—H8A \cdots O5 ⁱ	0.85	1.82	2.643 (2)	163
O8—H8B \cdots O10 ^j	0.85	1.90	2.744 (2)	176
O9—H9A \cdots O10 ⁱⁱⁱ	0.85	1.96	2.786 (2)	164
O9—H9B \cdots O11 ^{iv}	0.85	1.92	2.763 (2)	173
O10—H10A \cdots O3 ^{iv}	0.85	1.89	2.731 (2)	172
O10—H10B \cdots O2 ^{iv}	0.85	2.05	2.886 (2)	169
O11—H11A \cdots N6 ⁱ	0.85	2.04	2.868 (2)	165
O11—H11B \cdots N5 ^v	0.85	2.05	2.894 (2)	174

Symmetry codes: (i) x, y, z ; (ii) $x, y + 1, z$; (iii) $x, y - 1, z$; (iv) $x - 1, y, z$; (v) $-x + 2, -y + 1, -z + 1$.

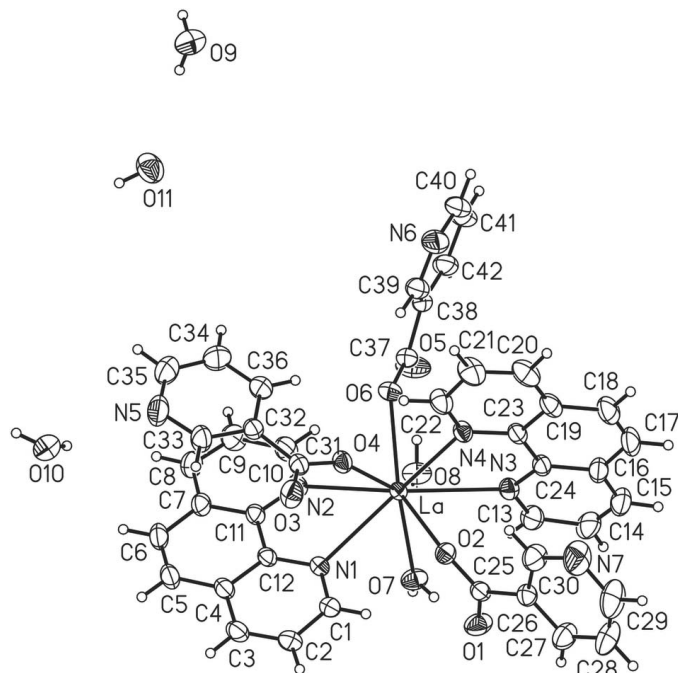


Figure 1

The asymmetric unit of (I), showing 35% probability displacement ellipsoids (arbitrary spheres for H atoms).

H atoms attached to C atoms were included at calculated positions and treated as riding atoms [$C\text{---}H = 0.93$ Å and $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$]. The water H atoms were found in a difference map, relocated in idealized positions ($O\text{---}H = 0.85$ Å) and refined as riding with $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{O})$.

Data collection: SMART (Bruker, 1998); cell refinement: SMART; data reduction: SAINT (Bruker, 1998); program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: SHELXTL (Bruker, 1998); software used to prepare material for publication: SHELXTL.

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